

HPLC method development for ibuprofen quantification

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#### Project

Develop the HPLC method to quantify ibuprofen and its release from the formulation



# Outline of my presentation

- Method development and optimization
- Method validation
- Sample preparation and application of the HPLC method



I am sorry that I only have this blurry picture for this chromatogram as the computer was infected with a virus and everything was erased from the computer.



# Method development

Stationary phase	Mobile phase	Chromatographic time (mins)	Flow pH rate(mL/mi n)		Detection
CLC Shim-pack C8 column (2504.6 mm, 5-Im particle size)	methanol and 0.05M dihydrogen phosphate buffer (75:25, v=v)	7	6.5	1	fluorescence- detection at 295nm
Chiracel OJ-RH column (150 × 2.1 mm, 5 µm)	mixture of 0.1% (v/v) acetic acid in methanol/water (90:10, v/v)	6	/	0.15	MSMS
(R, R)-Whelk-02 (4.6 mm diameterx 250 mm leng th, I 0 ~Im partic le size	EtOH-water (30 + 70 v/v) containing I 00 mtvl AmAc	12	/	1.3	diode array detector 220nm
LuxCellulose 3 (250 × 4.6 mm, 5 m particle size)	0.1% (v/v) acetic acid in mixture of methanol and water in ratio90:10	14		0.6	MS
Agilent ZORBAX Eclipse Plus C18 100 mm × 4.6 mm,3.5 mm column	methanol	4		/	DAD 221nm
Luna® 5 µm C18(2) 100 Å, LC Column 250 x 4.6 mm, Ea	Acetonitrile / Water (60:40) with 20mM Chloroacetic acid	6	3	2	254 nm
C18 column (Hypersil BDS, 150 x 4.6 mm, 5 µm)	Acetate buffer (triethylamine & ortho phosphoric acid ) andACN in the ratio of 40:60 % (v/v).	4		1.5	220 nm





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# Method optimization





## Chromatogram

- Run time: less than 7 minutes
- Sharp peaks



#### Method validation

Batch:



Run







# Results (1)





# Results (2)

	Accuracy (%)					Precisj <del>on (%)</del>					
Concentration	Within-run		Between-	Acceptance		Within-ru	un	Between-	Acceptance		
(ug/mL)	1	2	3	run	criteria +/-	1	2	3	run	criteria +/-	
2	114	78	101	100	20	9	2	7	20	20	
5	98	99	97	98		8	4	7	6		
20	96	103	105	93	10	10	1	6	4	6	10
50	96	100	97	91		8	6	4	5		

The linearity, accuracy and precision of the method were successfully validated within the concentration range of 5-50 ug/mL.

### Sample preparation





#### Results



Sample 1

Sample 4



## Next steps

- Robustness of the method
- Method suitability test



## Acknowledgement

- Thank you Syaza, Natalie, Given and Gracia
- This project would not have been possible without your support and guidance.



#### Calculations

	Concentration (mg/mL)	Stock solution (mL)	ACN (mL)	Total volume (mL)	
Ibuprofen stock solution	10				
Diluted solution A	1	0.1	0.9	1	
Diluted solution B	0.1	0.1	0.9	1	
Ketoprofen stock solution	10				
Diluted solution C	1	0.1	0.9	1	
Diluted solution D	0.1	0.1	0.9	1	
	Usin	g diluted solution B ar	nd D:		
	Concentration (ug/mL)	Solution B (ibu) (uL)	ACN (uL)	Solution D (keto) (uL)	Total volume (mL)
Standard solution 1	2	20	978	2	1
Standard solution 2	5	50	948	2	1
Standard solution 3	10	100	898	2	1
Standard solution 4	20	200	798	2	1
Standard solution 5	30	300	698	2	1
Standard solution 6	40	400	598	2	1
Standard solution 7	50	500	498	2	1
Standard solution 8	2.5	25	973	2	1



#### DOE

1	1							
		Effect of	Effect of	Effect of	Effect of	Effect of	Effect of	Effect of
		factor 1	factor 2	factor 3	interaction	interaction	interaction	interaction
level	average	Tween 80	sodium citrate	sodium alginate	IA12	IA13	IA23	IA123
level -1		0	0	150				
level 0		50	150	200				
level +1		100	300	250				
unit	in 10 mL	mg	mg	mg				
exp1	1	-1	-1	-1	1	1	1	-1
exp2	1	1	-1	-1	-1	-1	1	1
exp3	1	-1	1	-1	-1	1	-1	1
exp4	1	1	1	-1	1	-1	-1	-1
exp5	1	-1	-1	1	1	-1	-1	1
exp6	1	1	-1	1	-1	1	-1	-1
exp7	1	-1	1	1	-1	-1	1	-1
exp8	1	1	1	1	1	1	1	1
CP1		0	0	0	0	0	0	0
CP2		0	0	0	0	0	0	0
CP3		0	0	0	0	0	0	0



# Calculation (sample prep)

- Concentration of ibu in the formulation is 20 mg/10 mL (2 mg/mL)
- Test of release of the formulation: 600 ug/mL to 3 mL of HCI (0.4 mg/mL)
- Dilution method: assumed all ibu is released,
- Make it up to '30 ug/mL'
- 74 uL sample solution was used (1:13.3 dilution)
- we don't want it to go over the concentration of 50 ug/mL which is the limit of the validated calibration graph.



#### Use of internal standard?





#### Wavelength







# Composition of formulation

 The solution is prepared by mixing sodium alginate, T80, calcium carbonate and sodium citrate.

HCI



## May I remove outliers?

If an outlying value can be traced back to a failure in the system (e.g. injection error, bad chromatography, pipetting error, etc.) then it is permissible to remove it or better yet to repeat the measurement in question. If such a retrace does not come up with any failure then the outlying value should be considered as a real but rare incident and kept in the data.



#### How it treats JA



#### Bracketing QCs

#### Batch:



Run



#### Increase in temperature

#### **Baseline drift**

Possible cause	Solution
Column temperature fluctuation. Even small changes cause cyclic baseline rise and fall. Most often affects refractive index, UV and conductivity detectors at high sensitivity	Control column and mobile phase temperature, use heat exchanger before detector



Concentration	Day 1	Day 2	Day 3	average	sd	%accuracy	%precision	
2	2.186	1.516	2.074	1.99183333	0.4030391	100%	20%	
2	2.765	1.573	1.933					
2	2.53	1.561	2.195					
2	2.142	1.563	1.864					
5	4.672	4.658	4.544	4.89625	0.30462052	98%	6%	
5	4.75	5.149	5.091					
5	4.721	4.921	5.15					
5	5.531	5.051	4.517					
20	18.977	22.29	20.811	18.6695	5.98521219	93%	32%	
20	0	20.676	22.184					
20	19.183	19.757	20.112					
20	19.292	19.483	21.269					
50	44.025	48.527	47.01	45.599	11.6255877	91%	25%	
50	48.954	54.339	50.601					
50	51.129	48.862	46.886					
50	9.568	47.826	49.461					